

Seminar Series -24 February

2025

NETCASTPL4.0

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NetCastPL4.0 Seminars Series – 24 February 2025

Today's seminar presentations are focused on **Green Molding technologies**, specifically inorganic foundry binders for sand molding processes

Presentations

- State-of-the-art and challenges in inorganic binder systems
- Potential of solid inorganic foundry binders

Questions can be added during presentations through the QA function



State-of-the-art and challenges in inorganic binder systems

Kalle Jalava

Background

Foundries are major players in metals industry

Negative environmental effects of foundries result from thermal processes and the use of organic and mineral additives



Background

In the making of moulds and cores various additives are used to bind sand

In the binding of sand and pouring of metal, **reaction and decomposition products are generated**

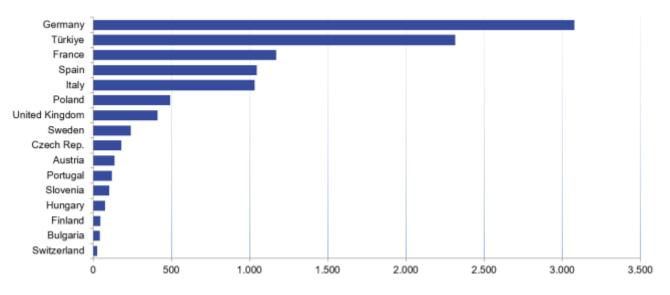


Background

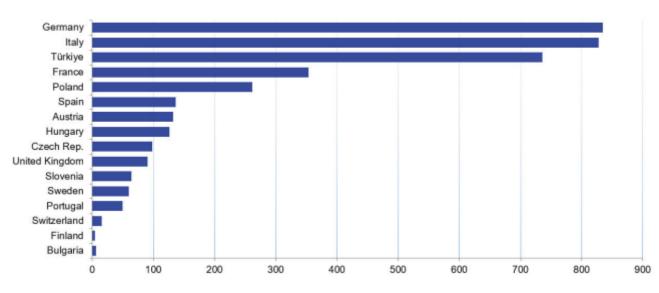
Production of Iron, Ductile Iron and Steel Castings in the European Foundry Industry 2023 (in 1.000 t)

Inorganic binder systems can be an answer for the future needs for sustainable manufacturing of ferrous castings

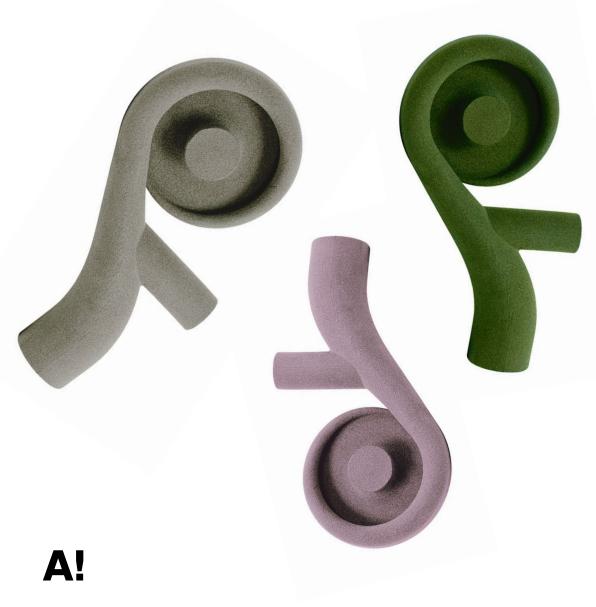
...while certain technological challenges remain to be solved



Production of Non-Ferrous Metal Castings in the European Foundry Industry 2023 (in 1.000 t)



Foundry binders – to name a few



Organic

Furan, phenolic, …

Inorganic

- Sodium silicates
- Geopolymers

Semi-inorganic

• Chemically cured silicates and geopolymers (aluminosilicates)

Why the need for change?

A lot of the used organic resin binder systems are classified as **harmful or hazardous**

Even though the application properties are **good, tried and tested**, it will be increasingly harder to comply with future regulations In foundries, hazardous substances and substances of very high concern which are used in moulding and core-making and which are listed in Regulation (EC) No 1272/2008 include, for example:

- *Furan*: Classified as carcinogen (Category 1B H350: 'May cause cancer').
- Phenol: Classified as mutagen (Category 2 H341: 'Suspected of causing genetic defects').
- *Methylene diphenyl diisocyanate (MDI):* Classified as carcinogen (Category 2 H351: 'Suspected to be carcinogenic').
- *Formaldehyde:* Classified as carcinogen (Category 1B H350: 'May cause cancer') and mutagen (Category 2 H341: 'Suspected of causing genetic defects').
- *Furfuryl alcohol:* Classified as carcinogen (Category 2 H351: 'Suspected to be carcinogenic').
- *Benzene:* Classified as carcinogen (Category 1B H350: 'May cause cancer') and mutagen (Category 1B H340: 'Germ cell mutagenicity').

EU SF BREF

Why the need for change?

Upcoming and future regulation on the emissions for foundries will limit the use of the industrial standard organic binder systems

Table 4.9:BAT-associated emission levels (BAT-AELs) for channelled emissions to air of dust,
benzene, formaldehyde, phenol and TVOC from casting, cooling and shake-out
processes in foundries using lost moulds including the full mould process

Substance/Parameter	Unit	BAT-AEL (Daily average or average over the sampling period)			
Dust		1 - 5			
Benzene	mg/Nm ³	< 1 - 2 (1)			
Formaldehyde		$< 1 - 2 (^2)$			
Phenol		$< 1 - 2 (^{3})$			
TVOC	mg C/Nm ³	15 – 50 (⁴)			
 (¹) The BAT-AEL only applies when aromatic binders/chemicals are used or when the full mould process is used. (²) The BAT-AEL only applies when the substance concerned is identified as relevant in the waste gas streams based on the inventory of inputs and outputs mentioned in BAT 2. (³) The BAT-AEL only applies when phenolic-based binder systems are used in moulding and/or core-making. (⁴) The upper end of the BAT-AEL range may be higher and up to 100 mg C/Nm³ when organic binder systems generating low or no emissions of substances classified as CMR 1A, CMR 1B or CMR 2 (see techniques (d), (e) and/or (f) in BAT 25) are used in core-making. 					

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The potential

The generated harmful emissions from inorganic and semi-organic systems are a fraction of the fully organic systems

Code	Per 1g of binder, [µg]		Per 1kg of molding sand, [mg]	
	Total BTEX	Benzene	Total BTEX	Benzene
MF	43,852	40,158	658	602 Furan
MA	32,994	30,911	495	464 Phenolic
MA/MF	0.752	0.770	0.752	0.770
MG	3342	2837	60	51 Geopol. + est
MG/MF	0.076	0.071	0.091	0.085
мс	715	496	24	16 Sodium silicat
MC/MF	0.016	0.012	0.036	0.026
MB	2510	2301	176	161 Greensand
MB/MF	0.057	0.057	0.267	0.267
МІ	860	556	22	14 Sodium silicat
MI/MF	0.020	0.014	0.033	0.023

R. Danko et al.

Inorganic binder systems

The good

The bad

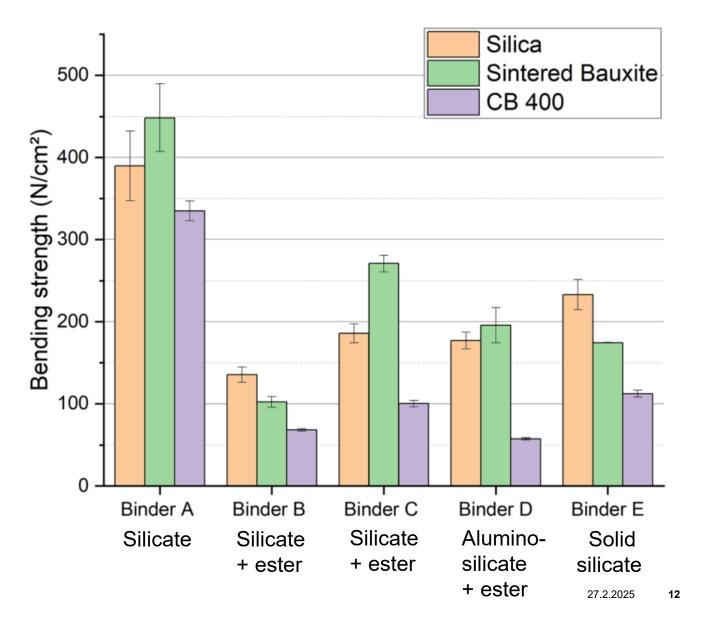
Little to no emissions Work well for non-ferrous castings Additional hardening at elevated temperatures, leading to challenges with **ferrous castings**

Inorganic binders

Inorganic binders generally achieve **high strength**

Semi-organic systems do not reach exactly the same levels

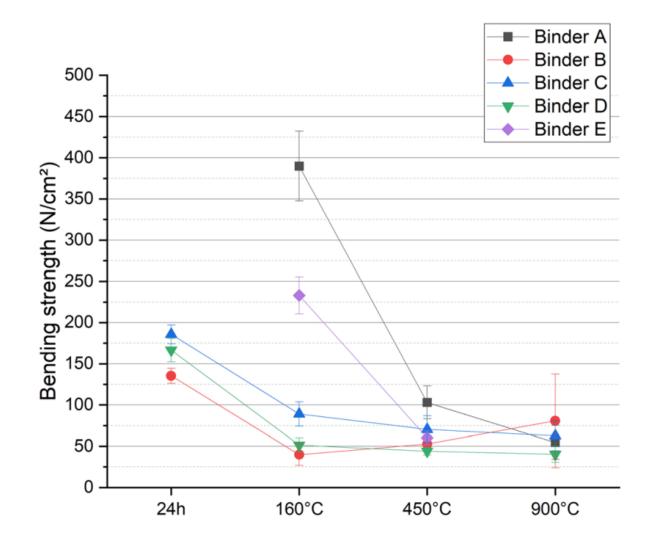
Modifiers/additives are used to customize properties



Challenges

Organic binders exhibit great collapsibility due to **thermal degradation**

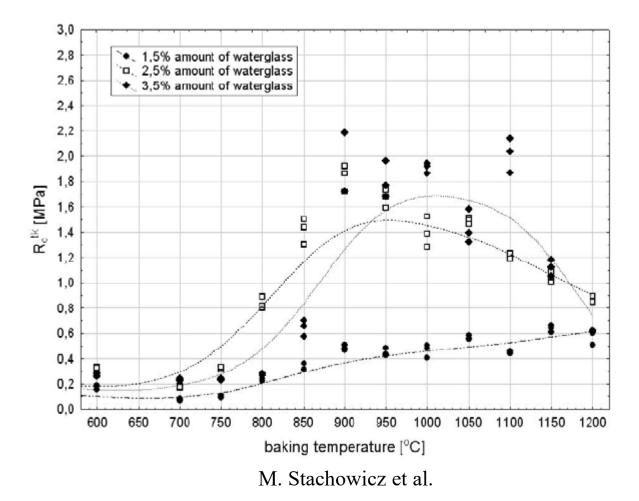
Inorganic binders do not exhibit this same behavior, leading to challenges with **ferrous castings**



Challenges

Organic binders exhibit great collapsibility due to **thermal degradation**

Inorganic binders do not exhibit this same behavior, leading to challenges with **ferrous castings**



Inorganic binder development in the industry

Modified sodium silicates

Chemically hardened sodium silicates

Geopolymers

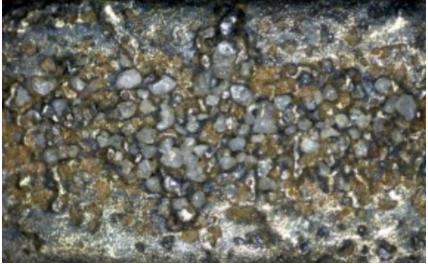
The solution to all challenges?



Fully inorganic systems will naturally have **the lowest emissions** ...while they are technologically hardest to implement Semi-organic systems (geopolymers and silicates with hardeners) have potential in replacing **no-bake organic binder systems**

...ease of use

...collapsibility



Future needs for inorganic binder system use

Pattern technologies

Compatibility with fully inorganic systems?

Collapsibility

• Modification needs?

Mold coatings

Δ!

• Aid in collapsibility?

Sand reclamation

• Mechanical, wet?

